K-2

Maharashtra State Board of Technical Education, Mumbai

LABORATORY PLAN (LP)

Academic Year: 2025-26

Date: 09/12/2025

Institute Name & Code: K. K. Wagh Polytechnic, Nashik-3 (0078) Class: TYCH

Program and Code: Chemical Engineering (CH)

Course Index: CO606

Course Name: **PETROCHEMICAL TECHNOLOGY**Total Hrs: 45 Semester: 6th Scheme: K

Course Code & Abbr.: 316306(PCT)

Name of Faculty: Mrs. A. B. Shaikh

• INDUSTRY EXPECTED OUTCOME

Develop expertise in food processing, preservation techniques, oil-seed technology, additive use, and compliance with food laws and regulations in food processing industries.

• COURSE LEVEL LEARNING OUTCOMES (COS)

- CO606.1 Identify the components of petroleum.
- CO606.2 Identify fractions and its properties after fractionation of petroleum.
- CO606.3 Use the different refinery processes to enhance the properties of refinery fractions.
- CO606.4 Demonstrate the manufacturing of pure chemicals from C1 to C4 chemicals through flow chart.
- CO606.5 Use Udex process to separate aromatic hydrocarbons from reformate.

• Teaching and Examination Scheme:

				Learning Scheme			Learning Scheme							P	Assessn	nent So	cheme								
				C	Actual Contac rs/Wee	t			22	ation		The	eory		Ва		LL & T	SL	Base S	ed on SL	rks				
Course Code	Course Title	Abbr	Course Category/s	SLH SLH	SLH NLH		SLH NLH	SLH NLH	SLH SLH	SLH	SLH	NLH	Credits	er Dui	FA- TH	SA- TH	Tot	al	FA-	-PR	SA-	PR	SI	LA	Ma
				CL	TL	LL				Pap	Max	Max	Max	Min	Max	Min	Max	Min	Max	Min	Tota				
316306	PETROCHEMICAL TECHNOLOGY	РСТ	DSE	3	-	2	1	6	3	03	30	70	100	40	25	10	25#	10	25	10	175				

Abbreviations: CL- Class Room Learning, TL- Tutorial Learning, LL-Laboratory Learning, SLH-Self Learning Hours, NLH-Notional Learning Hours, FA - Formative Assessment, SA -Summative assessment, IKS - Indian Knowledge System, SLA - Self Learning Assessment

Legends: @ Internal Assessment, # External Assessment, *# On Line Examination, @\$ Internal Online Examination

• Laboratory Learning Outcome (LLO)

LLO No.	Title of LLO
LLO 1.1	Measure specific gravity of given sample using digital weight balance and then evaluate API gravity.
LLO 2.1	Measure the time required to collect 50 ml of sample. By using the time, calculate viscosity of given
	sample at different temperatures.
LLO 3.1	Measure the minimum temperature at which the vapours of given oil will give momentary flash and
	oil will burn at least for 5 seconds on the application of standard flame.
LLO 4.1	Measure the minimum temperature at which the vapours of given oil will give momentary flash and
	oil will burn at least for 5 seconds on the application of standard flame.
LLO 5.1	Measure the minimum temperature at which the vapours of given oil will give momentary flash and
	oil will burn at least for 5 seconds on the application of standard flame.
LLO 6.1	Measure the maximum height of the flame in millimeter at which the vapours of given oil will burn
	without giving smoke.
LLO 7.1	Measure the minimum temperature at which the equal volume of anhydrous aniline and given oil
	sample is completely miscible.
LLO 8.1	Measure the minimum temperature at which oil appears a cloudy nature and ceases to flow.

LLO 9.1	Measure the weight of carbon residue of given sample on digital balance by evaporating light
	components from it by means of heating.
LLO 10.1	Measure initial and final boiling point of given sample.
LLO 11.1	Measure the minimum temperature at which the wax melts and drop falls within test tube by
	providing paraffin liquid bath as a heating media.
LLO 12.1	After the completion of reaction, ethyl acetate layer is washed with water, filter it with filter paper
	and after drying then measure the yield of ethyl acetate.
LLO 13.1	After the completion of reaction, PF resin layer is washed with water, filter it with filter paper and
	after drying measure the yield of phenol formaldehyde resin.
LLO 14.1	Separate the layer of biodiesel and glycerine by means of separating funnel and then measure the
	volume of biodiesel layer.
LLO 15.1	Separate benzene-toluene liquid mixture using DWSIM chemical process simulator.
LLO 16.1	Measure the API gravity and aniline point of given sample and then calculate diesel index
	mathematically.

• COs, Practical Laboratory Learning Outcome (LLOs) and Mapping:

PR. No	Relevant COs	Practical - Laboratory Learning Outcome (LLO)	Practical Titles	Planned Dates		Actual Date of conduction	Remark/ Assess- ment Date with Staff sign
		()		From	То		
1	CO1 CO2	LLO 1.1	Determination of API gravity of given	A-19/12/25	A-26/12/25		
	CO4 CO5		petroleum sample using density bottle.	B-20/12/25	B-27/12/25		
2	CO1 CO2 LLO2.1		Determination of viscosity of given petroleum sample at	A-26/12/25	A-02/01/26		
	CO4 CO5	22021	different temperatures using Redwood viscometer.	B-27/12/25	B-03/01/26		
	G02	LLO 4.1	* Determination of the flash point and fire point of the given sample using Abel's apparatus.	A-02/01/26	A-09/01/26		
3	CO2			B-03/01/26	B-10/01/26		
4	CO2	LLO 6.1	* Determination of smoke point of given	A-09/01/26	A-16/01/26		
	4 CO2 LEO		sample.	B-10/01/26	B-17/01/26		
5	CO2	LLO 7.1	* Determination of aniline point of the	A-16/01/26	A-23/01/26		
	CO5		given sample.	B-17/01/26	B-24/01/26		
	CO2	LLO 8.1	* Determination of cloud point and pour	A-23/01/26	A-30/01/26		
6	CO3		point of the given sample.	B-24/01/26	B-31/01/26		
	CO1	11001	* Determination of carbon residue of a	A-30/01/26	A-06/02/26		
7	CO2	LLO 9.1	given sample using Conradson apparatus.	B-31/01/26	B-07/02/26		
8	CO2	LLO 11.1	* Determination of drop melting point of wax	A-06/02/26	A-13/02/26		

PR. No	Relevant COs	Practical - Laboratory Learning Outcome (LLO)	Practical Titles	Planned Dates		Actual Date of conduction	Remark/ Assess- ment Date with Staff sign
			using drop point apparatus.	B-07/02/26	B-14/02/26		
		LLO 12.1	Synthesis of ethyl acetate from acetic acid	A-13/02/26	A-20/02/26		
9	CO3		and ethyl alcohol using the esterification process.	B-14/02/26	B-21/02/26		
10	CO3	LLO 13.1	* Preparation of phenol	A-20/02/26	A-27/02/26		
10	CO3		formaldehyde resin.	B-21/02/26	B-28/02/26		
	CO4	LLO 14.1	* Preparation of biodiesel from used vegetable oil using a	A-27/02/26	A-06/03/26		
11	CO4		transesterification process.	B-28/02/26	B-07/03/26		
10	CO1	LLO 16.1	* Determination of the	A-06/03/26	A-13/03/26		
12	CO2 diesel i		diesel index of a given sample.	B-07/03/26	B-14/03/26		
13			Beyond Syllabus	A-13/03/26	A-20/03/26		
13			Practical.	B-14/03/26	B-21/03/26		

• ASSESSMENT METHODOLOGIES/TOOLS

A. Formative assessment (Assessment for Learning)

• Two Class Test of 30 Marks Each, Term Work Assessment of 25 Marks, and Self learning assessment of 25 marks.

B. Summative Assessment (Assessment of Learning)

• End Term Theory Examination, End Term Practical Examination

Laboratory Equipment / Instruments / Tools / Software required

Sr. No.	Equipment Name with Broad Specifications	Relevant LLO Number
1	Specific gravity bottle, capacity of 10ml / 25 ml.	1,16
2	ASTM distillation set up as per ASTM standard.	10
3	Drop point apparatus: Drop Point apparatus as per IP-132 and ASTM-D-566 20 to 120 °C x 1°C Or 100 to 230 °C x 1°C.	11
4	Conical flash (500 ml), beaker (500 ml), separating funnel, filter paper, weighing balance, dropper.	12,13,14
5	DWSIM Chemical process simulator	15
6	Redwood viscometer: electronic digital indicator and controller and FHP motor stirrer.	2
7	Pensky Marten Flash Point Apparatus: electric heater with temp, regulator. Suitable for operation on 220 Volts 50 cycles AC Circuits.	3
8	Abels Flash Point Apparatus: It is supplied with oil cup, cover fitted with stirrer, thermometer socket S.S.Water Bath, Stand. An electric heater is fitted at bottom for range -18°C to +70°C with Refrigeration System	4
9	Cleveland open cup flash point apparatus: as per ASTM standards	5
10	Smoke point apparatus: as per ASTM standards	6
11	Aniline Point Apparatus: Power: 10W, Supply Voltage: 220V/50Hz, 01 no.	7,16

	Transformer Unit Electrical heater with variable heat control, Multi-test verification material. Spares Kit, glassware, electrically heated with motorized stirrer	
12	Cloud & Pour Point Apparatus: glass bath jar, polished brass cylinder mounted on metal tripod base, glass test cylinder, cork bottom disc and top rings, thermometer (H-2600.5F -36° to 120°F) fitted into cork for sealing test cylinder.	8
13	Conradson Carbon Residue Apparatus: The apparatus consists of spun sheet iron crucible 25cc Capacity, Sheet Iron hood and sheet iron block on a stand with triangular wire, and gas Burner. Elect heating with separate regulator control.	9

• References:

• Suggested Learning Materials / Books:

Sr.	Author	Title of Book	Publication
1	B. K. Bhaskara Rao	Modern Petrochemical Refining processes	Oxford – IBH Publications, Delhi ISBN:9788120417115
2	B. K. Bhaskara Rao	A Text on Petrochemicals	Khanna Publishers, Delhi ISBN- 9788174090444
3	W. L. Nelson	Petroleum Refinery Engineering	McGraw Hill, New York Publications, ISBN: 9780070855366
4	Gary, James H Glenn E Handwork Mark J Kai sen	Petroleum Refining Technology and Economics	CRC Press, USA Publications, ISBN - 9780849370380
5	M. Gopal Rao and Sitting, Marshal	M. Gopal Rao and Sitting, Marshal	East-West Press Pvt. Ltd., Delhi Publications, ISBN- 9788185938790
6	Austin G.T.	Shreve's Chemical Process Industries	McGraw Hill India, Pune Publications, ISBN-9781259029455
7	Dr. Ram Prasad	Petroleum Refining Technology	Khanna Publishers, Delhi ISBN- 9788174090645
8	Sukumar Maiti	Introduction to Petrochemicals	Oxford - IBH Publications, Delhi ISBN:9788120406636
9	James G. Speight	Handbook of petrochemical processes	CRC Press Publications, ISBN: 10: 1498729703

• Learning Websites & Portal

Sr. No	Link / Portal	Description
1	https://onlinecourses.nptel.ac.in/noc23_ch64/preview https://nptel.ac.in/courses/103105221 https://nptel.ac.in/courses/114106042 https://www.youtube.com/watch?v=Js4cxZRtk5Q https://archive.nptel.ac.in/courses/103/102/103102022/ https://archive.nptel.ac.in/content/syllabus_pdf/103102022.p df http://digimat.in/nptel/courses/video/103107081/L15.html http://kcl.digimat.in/nptel/courses/video/103103220/L19.html https://www.youtube.com/watch?v=az-q9ga5M0g https://www.youtube.com/watch?v=JAZTQhIoEd8 https://www.youtube.com/watch?v=CZCdDpm3SmE https://www.youtube.com/watch?v=LpWRMYgSatw https://archive.nptel.ac.in/courses/103/107/103107082/ https://archive.nptel.ac.in/courses/103/103/103103217/ https://archive.nptel.ac.in/courses/103/107/103107082/ https://archive.nptel.ac.in/courses/103/107/103107082/ https://archive.nptel.ac.in/courses/103/107/103107082/ https://archive.nptel.ac.in/courses/video/103107081/L27.html http://acl.digimat.in/nptel/courses/video/103107212/L58.html https://www.youtube.com/watch?v=s1N_VhgbWv4	Petrochemical Technology

Mrs.A. B. Shaikh (Name & signature of staff)

Dr. P. S. Bhandari (Name & signature of HOD)